

# 2.3.1

**Student centric methods** 

(Experimental learning, Participative learning and Problem solving methodology)

A) Practical Activity SessionsB) Practical Activity Sessions (Sample copy)

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**Student centric methods** 

(Experimental learning, Participative learning and Problem solving methodology)

**A)Practical Activity Session** 

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# 2.3.1

**Student centric methods** 

A) Practical Activity Session

1) Department of Pharmacognosy





#### Anand Pharmacy College, Anand Department of Pharmacognosy PRACTICAL ACTIVITY (LEARN WITH FUN) Academic Year: 2022-23

#### Report on the activity done during the academic year

B.PHARM			
ODD SEMI	ESTER		Defendent
Semester	Subject	Activity Planned	Activity Performed
3	Pharmacognosy&Phytochemistry-	13 ( in 4 batches)	17
5	Pharmacognosy&Phytochemistry- 2	13 (in 4 batches)	17
EVEN SEM	ESTER		
Semester	Subject	Activity Planned	Activity Performed
6	Herbal Drug Technology	6	15

**NOTE:** Deviation was observed from planned and performed activities due to the length of the practical, explanation, calculation, and understanding, of the practical task. This made the time constraints, so it was difficult to conduct activities due to lack of time.

Prepared & submitted by:

**Dr. Geetha Madhira,** Head of the department, Dept. of Pharmacognosy Anand Pharmacy College, Anand.







#### Anand Pharmacy College, Anand Department of Pharmacognosy PRACTICAL ACTIVITY (LEARN WITH FUN) Academic Year: 2022-23

Report on the activity done during the academic year

PHARM D Year	Subject	Activity Planned	Activity Performed
2	Pharmacognosy & Phytocheuticlas	21 (in 2 batches)	16 (in 2 batches)

**NOTE:** Deviation was observed from planned and performed activities due to the length of the practical, explanation, calculation, and understanding, of the practical task. This made the time constraints, so it was difficult to conduct activities due to lack of time.

Prepared & submitted by:

**Dr. Geetha Madhira,** Head of the department, Dept. of Pharmacognosy Anand Pharmacy College, Anand.

**Mr. Subhashchandra K Patel** Associate Professor, Dept. of Pharmacognosy Anand Pharmacy College, Anand.

B. Pharm/Pharm D: GHL Subject name: HDT Batch: A Name of Faculty: Dr. GEETHA Name of Activity: Prepara Description of the Activity:	Anand Pharmacy College, Ana Academic Year: 2022-23 Department of Pharmacognosy PRACTICAL ACTIVITY FEEDBACK (LEARN WITH FUN)	Semester/Year: 22-21 Subject code: 13P603TP Date: 1714122
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Prepared by:

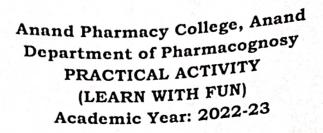
GUE)+A.M Faculty Name: Mr. Subhashchandra-K-Patel Designation: Associate Professor Dept. of Pharmacognosy

Checked by:

Nex Dr. Geetha Madhira HOD (B. Pharm & Pharm D) Dept. of Pharmacognosy a this repo 1.12

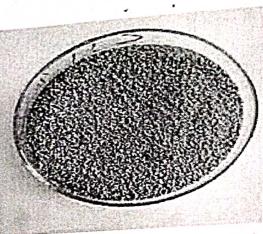














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De	nd Pharmacy College, Anar Academic Year: 2022-23 partment of Pharmacognosy ACTICAL ACTIVITY FEEDBACK (LEARN WITH FUN) and Phytochemistry-II 4 for find the topic	Semester/Year: Subject code: B	2
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Prepared by:

Faculty Name: Janvi P. Chauhan Designation: Assistant Professor Checked by:

Dr. Geetha Madhira HOD (B. Pharm & Pharm D) Dept. of Pharmacognosy

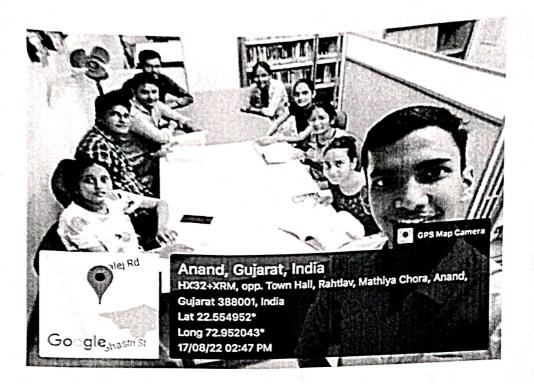
Dept. of Pharmaceutical chemistry Note: Attach photographs (maximum 2) with this report and submit final hard copy



Anand Pharmacy College, Anand Academic Year: 2022-23 Department of Pharmacognosy PRACTICAL ACTIVITY FEEDBACK (LEARN WITH FUN)



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	nd Pharmacy College, Anar Academic Year: 2022-23 partment of Pharmacognosy ACTICAL ACTIVITY FEEDBACK	
B. Pharm: B. Pharm	(LEARS, WE	Semester/Year: 3 <sup>rd</sup> Sem, 22-23 Subject code: BP305TP Date: 1011012022
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Batch: D Name of Faculty: Sneha Pare Name of Activity:. Video C Description of the Activity: Stude Name of students involved	Students feedback	e cohar 15 Signature Chadans
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Prepared by:

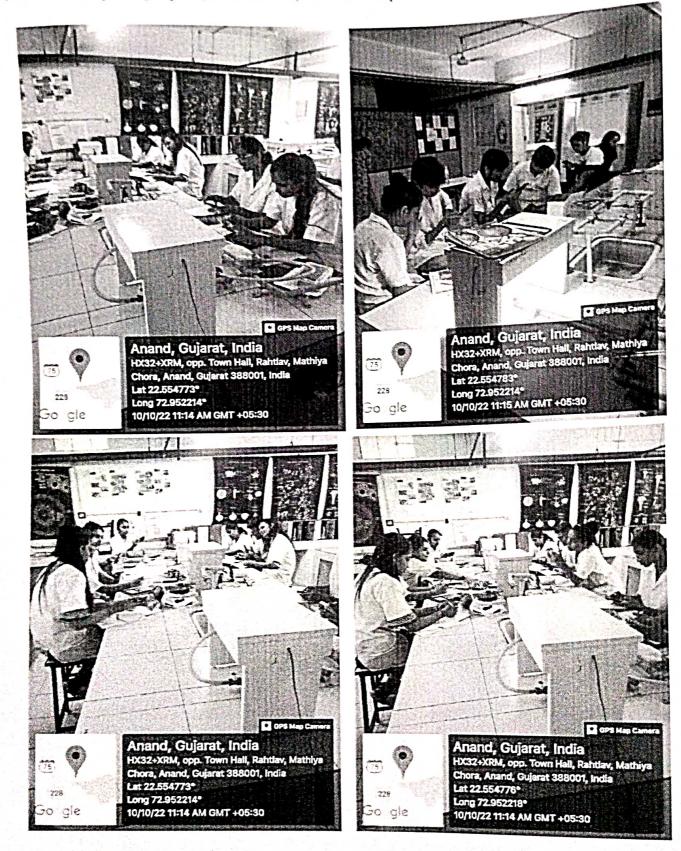
Smiller Faculty Name: Snehel Peutel (SRP) Designation: Assistent Professor Dept. of Pharmacognosy

J)

Dr. Geetha Madhira HOD (B. Pharm & Pharm D) Dept. of Pharmacognosy Anand Pharmacy College, Anand Academic Year: 2022-23 Department of Pharmacognosy PRACTICAL ACTIVITY FEEDBACK (LEARN WITH FUN)



Note: Attach photographs (maximum 2) with this report and submit final hard copy



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# 2.3.1 Student centric methods A) Practical Activity Session 2) Department of Pharmaceutics



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Name of Faculty: Mitak Oak	1	Date: 28/11/22
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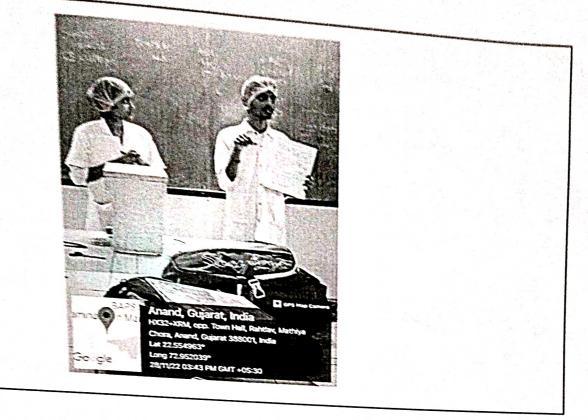
Checked by:

Paculty Name Milas Pales Designation Dept. of Pharmaceutics

Dr. Dipika Chavda HOD (B.Pharm & PharmD) on Pharmaceutics Dept. of Pharmaceutics Note: Attack photographs (maximum 2) with this report and submit final hard copy



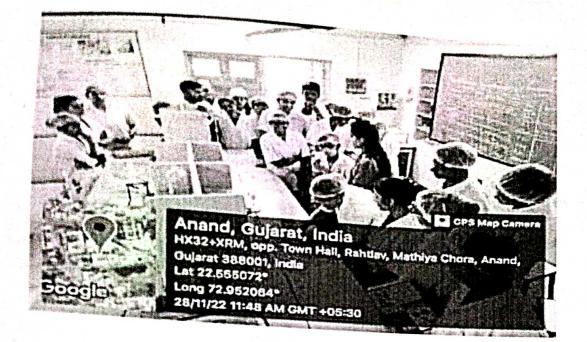
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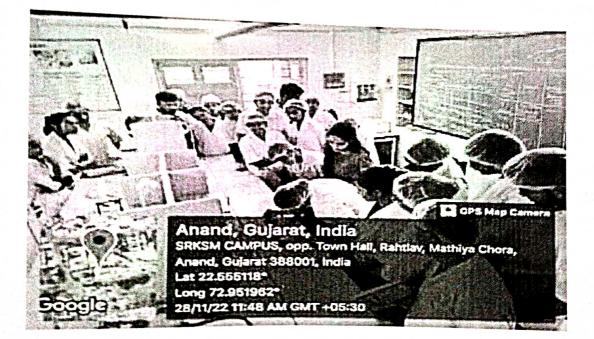






Anand Pharmacy College, Anand Academic Year: 2022-23 Department of Pharmaceutics PRACTICAL ACTIVITY FEEDBACK (LEARN WITH FUN) B.Pharm/PharmD: B.Phunm Subject name: Physical Phanma ceutico I Semester/Year: 300 Batch: B Subject code: BP302TP Name of Faculty: Ms. Ayoshi Patel Date: 28/22/2022 Name of Activity: Related to Surface tension & Solvens. Name of students involved Students feedback (Ioswami. Mana Signature Excellent Vhora ana Dar (500c 919 ar Xcellent lai ratham Very Grooc 1 ci Kuni Very 6400d andan GOOD Excellent tin v21. Magical, Good Vh Nathren Mayical VERY GOOD Much Better. MMI D. P. Paty Good But Improvable mavsa shyshi liuati Aussome Ninguttin Damn 4000 Magnificient Meher Vama Amazing Chudasamor Khavna - Mend Blowling Prepared by: Checked by: Faculty Name Ayoshi Patel Dr. Dipika Chavda Designation Asid Professor. HOD (B.Pharm & PharmD) Dept. of Pharmaceutics Dept. of Pharmaceutics Note: Attach photographs (maximum 2) with this report and submit final hard copy





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Anand Pharmacy College, Anand Academic Year: 2022-23 Department of Pharmaceutics PRACTICAL ACTIVITY FEEDBACK (LEARN WITH FUN)



B.Pharm/PharmD: Subject name: Pharmacentics - I Batch: B. Name of Faculty: Dr. Chetna Modi

Semester/Year: First BP Subject code: BP103TP Date: 18/11/22\_

Name of Activity: Latin to	alk	
Description of the Activity: Senta	ences given to stude to a	in and that
Name of students involved	students feedback	Jeruns Signature
Isha Ka, patel	very informative.	J.K.K
Kashish. Shah	very interesting	KESTSh .
Patel Jiya	informative	10a
Patel vaide hi	informative	V.N. Pate
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Mauli Bhatt	Very informative	M-H-Bhatt
Janvi Pazeleh	Informative	J. Paroleh
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Desai Anyan	Informative	Qui
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Rana Hanswajsinh		
Sojitay Dhyuu	Got exposure to Latin coords	Detising

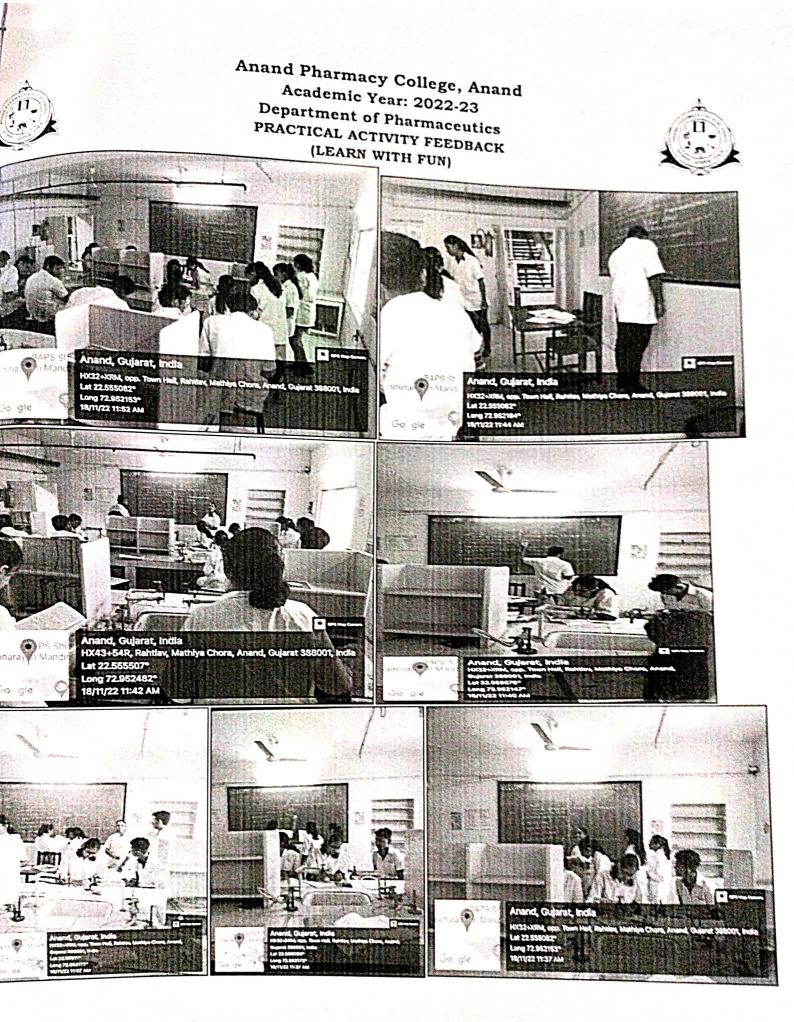
Prepared by:

Faculty Name Dr. Chetna Mod Professor.

Designation Dept. of Pharmaceutics Checked by:

Dr. Dipika Chavda HOD (B.Pharm & PharmD) Dept. of Pharmaceutics

Note: Attach photographs (maximum 2) with this report and submit final hard copy



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# 2.3.1

### **Student centric methods**

- A) Practical Activity Session
- **3)** Department of Pharmaceutical Chemistry

Anand Pharmacy College, Anand Academic Year: 2022-23 Department of Pharmaceutical Chemistry ACTIVITY

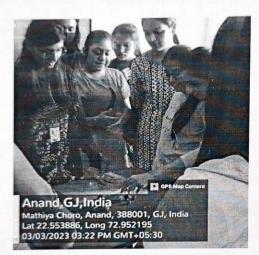
#### Course: B.Pharm (Class I and II)

Semester: IV

Date: 03.03.2023

Subject/code: Pharmaceutical Organic Chemistry (BP202TP) Name of Activity: Stereochemistry with Ball and Stick model

Objectives: To know how to learn stereochemistry with ball and stick model







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1: knowledge gain, 2- Problem solving capability, 3-Time management, 4- Team Work

5- Decision making, 6- Communication skill, 7- Creativity

Prepared by 000. gopi s. patel 2

Name & sign: Dept. of Pharmaceutical Chemistry Checked by:

Name & sign: HOD Dept. of Pharmaceutical Chemistry



Anand Pharmacy College, Anand Academic Year: 2022-23 Department of Pharmaceutical Chemistry ACTIVITY FEEDBACK



Sr. no.	Name of students	1	2	3	4	5	6	7	Sign	Remarks
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18	Tany Kyshwahu	V	V		V		V	~	Dustan.	a de la caractería
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21	River Partel	V	~	V		~	V	~	Put	
22	Klaushi Patel	L	V	V		V	V	V	1 Januar	
23	Tanisha Patel	V	U	V	-	V	~	~	Tandar	
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1: knowledge gain, 2- Problem solving capability, 3-Time management, 4- Team Work 5- Decision making, 6- Communication skill, 7- Creativity

Prepared b

DO. CAUPI S. Patel

Name & sign: Dept. of Pharmaceutical Chemistry

Checked by:

Name & sign: HOD Dept. of Pharmaceutical Chemistry

Anand Pharmacy College, Anand Academic Year: 2022-23 Department of Pharmaceutical Chemistry PRACTICAL ACTIVITY

Course: B. Pharm

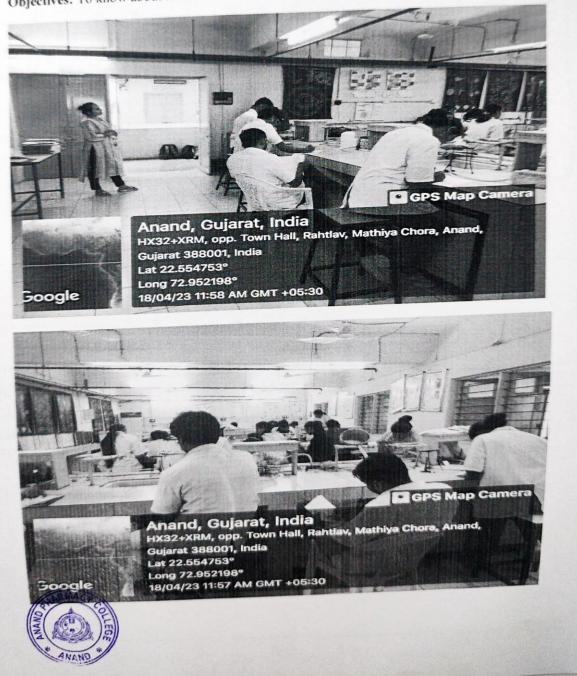
Semester: 4th (batch B)

Subject/code: Medicinal chemistry (BP402TP)

Date: 18.04.23

Name of Activity: Puzzle Solving

Objectives: To know about team work and knowledge gain



Descript	name and code: Poc activity: Match ion of the activity: To	CBP2.	B.Pho 02 TP. foll	j owin	g ymd		Semes Batch	: B	6-23 ar: 2nd	
Sr. no.	Name of students	1	2	3	4	5	6	7	Sign	Remark
1	Malek Tarannum	V							T.S.Mar	
2	Patel Banson	V							@patel:	
3	Janvi Parekh	$\checkmark$							& Paster	7
4	Dhara Pazekh	V							Phasa	
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7	Mansuzi Schein	V							Schein	
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1: knowledge gain, 2- Problem solving capability, 3-Time management, 4- Team Work

5- Decision making, 6- Communication skill, 7- Creativity

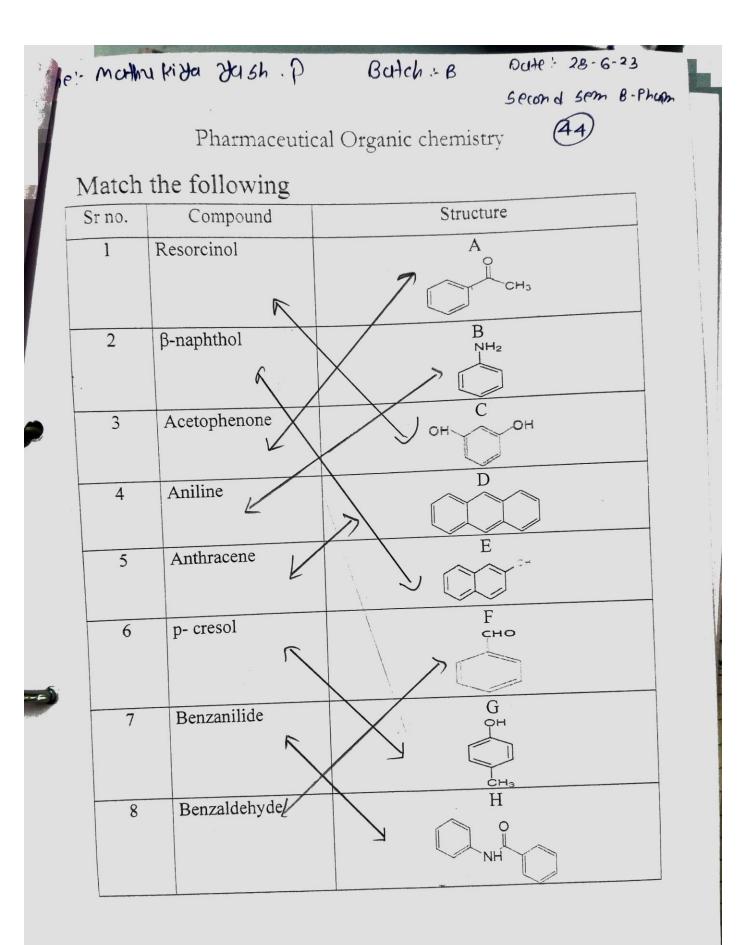
Prepared by:

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- Joc oft Name & sign: Dept. of Pharmaceutical Chemistry

Name & sign: HOD Dept. of Pharmaceutical Chemistry

....aceutical Chemistry



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	ion of the activity: To			bout	Knou	a for	Batch	"B yeilr	Roper	Hies P	
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1: knowledge gain, 2- Problem solving capability, 3-Time management, 4- Team Work 5- Decision making, 6- Communication skill, 7- Creativity

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Checked by:

Name & sign: HOD Dept. of Pharmaceutical Chemistry

	Anand Pharmacy College, Anand Academic Year: 2022-23 Department of Pharmaceutical Chemistry PRACTICAL ACTIVITY FEEDBACK										
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1: knowledge gain, 2- Problem solving capability, 3-Time management, 4- Team Work

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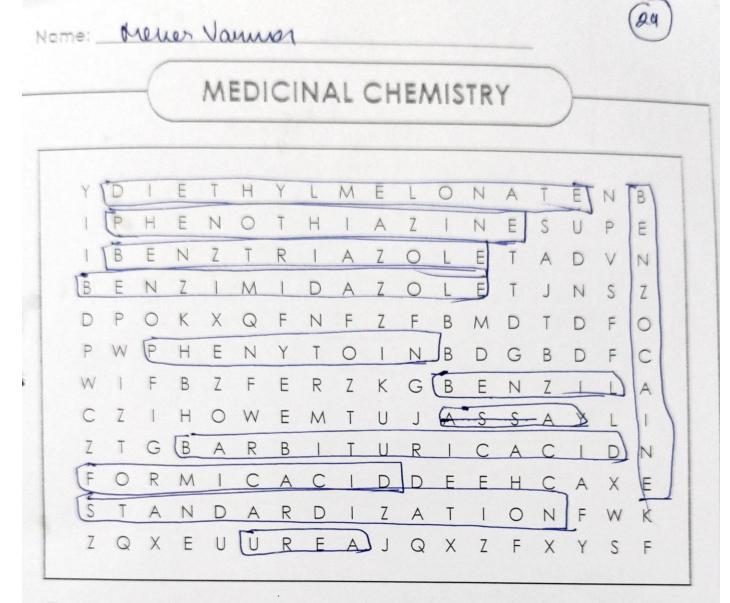
Name & sign: Dept. of Pharmaceutical Chemistry



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Name & sign: HOD Dept. of Pharmaceutical Chemistry



Find the following words in the puzzle. Words are hidden  $\rightarrow \psi$  and  $\checkmark$ .

ASSAY BARBITURICACID BENZIL BENZIMIDAZOLE BENZOCAINE BENZTRIAZOLE DIETHYLMELONATE FORMICACID PHENOTHIAZINE PHENYTOIN

-STANDARDIZATION



NOVEL TEACHING STRATEGIES TO IMPROVE STUDENT PERFORMANCE

## Pharmaceutical Analysis –I 1st semester B.Pharm

#### Objective

It is often difficult to engage Pharmacy students and keep their interest when teaching pharmaceutical analysis. These modules are designed for an effective change from a lecture-only approach to teaching with the use of games and case studies in a baccalaureate Pharmacy education program.

## SOP for Activity

# Pharmaceutical Analysis (1<sup>st</sup> sem B.Pharm)

## Index

Practical no	Activity	Page
		no
2	Copeia Puzzle	1
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6	Match making	3
7	Odd person	4
8	Memory Nomads	5
9	Role shift	6
11-15	Simul ad gradum	7
16	Group acitivity	8

Aim: To perform calibration of volumetric apparatus (pipette, burette and volumetric flask).

#### Objective

To achieve knowledge about calibration, its importance and to learn the methods to calibrate the volumetric apparatus.

#### Activity: Copoeia Puzzle

Copoeia Puzzle – About IP, calibration etc.

Students will be divided in four groups(in each group 4 to students) they will be given 20 minutes time to prepare the questions (with or without options) and each team will then cross question the other team, the team that succeeds to give the maximum answer will be declared winner.

**Outcome**: Students will update and enhance their knowledge regarding IP and Calibration.

Aim: To prepare and standardize 0.1 N Hydrochloric acid solution.

#### Objective

To achieve knowledge about preparation of HCl and to gain knowledge about how to prepare accurate solution in analysis

#### **Activity: Thinking Boats**

• The topic (Learning Definitions of molarity, molality, formality, normality, calculations) will be explained, the groups will be given the task the team that gives the fastest correct answer (Solved numerical) will be declared winner.

Outcome: Students will enhance their thinking ability and analytical ability.

**Aim**: To prepare and standardize 0.1 M Sodium thiosulphate solution as per IP 2014.

#### Objective

To achieve knowledge about preparation of Sodium thiosulphate.

#### Activity: Match making

Match the pair will be given and students will have to find the correct answer and will have to add a note on the matched pair. The group having the maximum correct answers with acceptable description will be announced winner.

**Outcome:** It will learn time management and decision making skill and that will help students to answer competitive exams.

AIM: To perform the assay of copper sulphate by iodometric titration.

#### Objective

To achieve knowledge about principle of assay.

#### Activity: Odd Person –

Series of words will be given, students will have to choose the odd one and will have to give the explanation about each even and odd chemical written.

**Outcome**: students will understand and update rationality of the prescription as a part of community pharmacist

**Aim**: To prepare and standardize of 0.02 M potassium permanganate solution as per IP 2014.

#### Objective

To achieve knowledge about preparation of potassium permanganate.

#### Activity: Memory Nomads -

"Permanganate titration" – will be taught to the groups and each group will prepare a question about the same, they will ask the fellow teams, the one giving correct answer will be announce winner.

**Outcome:** Students will update for group discussion.

**Aim**: To determine % w/v of hydrogen peroxide solution and volume strength of solution as per IP 2014.

#### Objective

To achieve knowledge about hydrogen peroxide, its advantage, disadvantage and calculation of strength.

#### Activity: Role Shift -

Learning calculation about all the performed practical, each group will learn one practical and will explain one practical. The group have the inputs from all the participating members and presenting acceptably will be declared winner.

Outcome: students will learn about calculation for particular practical.

## Practical No: 11-15

**Aim:** To determine the percentage purity (% w/w) of ferrous sulphate I.P.2014. (magnesium sulphate, calcium gluconate, Sodium benzoate, sodium chloride)

#### Objective

To achieve knowledge about principle, reaction, indicator.

Activity: Simul ad gradum – A step together to complete the story.

Each student will be designated a chemical name, he will prepare a drawing about what s/he is and will write it significance, so the first step require to complete the story (Reaction) will come and then the second has to step voluntarily, if s/he fails to do that, that is considered losing and then on that basis final winner will be declared, each team will work on this and then they will explain what they did, final winning will then be declared on the basis how sequentially they got arranged and how they explained.

**Outcome**: Students will learn the principal reaction, and also develop written and oral communication skills, confidence and minimize the stage fear.

#### **Practical No: 16**

**Aim**: To perform conductometric titration of hydrochloric acid with sodium hydroxide and to determine percentage purity of given acid (HCl).

## Objective

To achieve knowledge about conductometric principle, and conductometric titrations.

## Activity: Group discussion

- Student's batch will be dived into 4 groups
- Students will be given 4 different topic and they will discusses about each topic.

**Outcome**: It will be helpful to students to improve their verbal communication skill. Also they will learn to work in team and how to contribute in group discussion and leadership.



# 2.3.1

**Student centric methods** 

A) Practical Activity Session

4) Department of Pharmacology

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Sem- 5 Phamarologs . I D:03/8/22 (BPSO2TP) Batch= 0 Aim: Effect of daugs on isolated trog heart. Innovative Activity !-→ You-tube video of Fooluted frog hant experiment. (X-Phan Software) Question Answer of ditt daugs, effect on heart- and including Receptors action Mechanism pathway. 102 25 Visit to Library, Reter ditt booles of 2-2-25 elcology, = Outcomes Light = Understand the Rothe Dot effect of dang on heart, which Receptor gives action the where a with its mIA. 100 110 Roll No Signature 82 estandin 83 Bhanesterbert. Brakenuti 94 shivani 85 86 87 Absent 88 89 90

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Sem:-5 (Plate Batch D at X HEARY following Match the Types of PSS Frag Ringer (a) fot Uterus (i) Tyrode (b) Courinea Rig ileun (c) Friog heart tierre (d) Tracheal Persparahon (i) De Jalon (111) krebs solution (iv)(i)-c (ii)-b (iii)-a (jv)-d. N Roll no 1- Sign 04 Anjali pandey 10 2 Divya, Shray e 10 S 2 Fori 167 86 Ampads. 83 Rhang. 87 PasisamigerAB 98 Rohren 92 91 HARmehes 109 106 Herel proudy 89 88 Pepate 103 Funishy Asmina-J. Cl. Scienticiel 95 93 99

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